



Uncertainty analysis for calculations of the marine carbonate system for ICOS-Oceans stations

Tobias Steinhoff

NORCE Norwegian Research Centre AS, Bergen, Norway

tost@norceresearch.no

doi: 10.18160/vb7c-z758

ICOS OTC 2020

Contact Information:

ICOS OTC

Visiting address: Allegaten 70, NO-5007 Bergen, Norway

Postal address: Postbox 7810, NO-5020 Bergen, Norway

Email: icos-otc@uni.no

Motivation

Marine stations of the European infrastructure Integrated Carbon Observation System (ICOS) deliver high quality data to the Carbon Portal (CP). To ensure that every station fulfills the quality requirements of ICOS, they undergo a two-step labelling process. In the first step the station is evaluated on whether or not they can provide high quality data according to ICOS standards. In the second step the station must prove this by sending data and metadata to the Ocean Thematic Centre (OTC) showing that they measure data with the desired quality. The labelling scheme of ICOS-Oceans, as agreed during the Monitoring Station Assembly (MSA) meeting in Southampton in 2019, defines two goals with respect to the marine carbon cycle:

- Quantifying air-sea CO₂ fluxes
- Assessing the variability and drivers of these fluxes

Limitations of ocean *f*CO₂ measurement capabilities at Fixed Ocean Stations (FOS) mean that fluxes cannot be determined to the accuracy desired for large scale carbon budgets. Meanwhile, it is most common that Ship of Opportunity (SOOP lines) measure *f*CO₂ rather than other carbon system variables. Details about the labelling procedure can be found in the labelling document for ICOS marine stations, which is located on the OTC's website: <https://otc.icos-cp.eu/>. Stations are labelled as "Class 1" or "Class 2" (Table 1). Please note that there is no difference in quality between the two classes. A Class 1 station measures more variables than a Class 2 station.

Table 1: Measurement requirements for ICOS marine stations.

	Ship of Opportunity (SOOP) lines	Fixed Ocean Station (FOS)
Class 2 (minimum required variables)	<i>f</i> CO ₂ (±2 µatm)	<i>f</i> CO ₂ (±10 µatm) Alkalinity (TA) or Dissolved Inorganic Carbon (DIC) Oxygen
Class 1 (additional variables)	TA or DIC Oxygen	Surface: Nutrients (nitrate, silicate and phosphate)

For *f*CO₂ measurements onboard SOOP lines, the accuracy requirement for *f*CO₂ is ±2 µatm. If stations apply to be labelled as a Class 1 station, they need to provide one additional carbon variable (DIC or TA) and dissolved oxygen.

For FOS, the minimum requirement (Class 2) is that they measure *f*CO₂ with an accuracy of ±10 µatm and at least one additional variable of the carbonate system (DIC or TA) and dissolved oxygen.

This leads to two vital questions for the ICOS -Oceans stations:

A - How accurately should the second carbon variable be measured on SOOP lines and at FOS?

B - Can *f*CO₂ be calculated with a sufficient uncertainty by two other carbon parameters?

Knowing two of the four carbonate system variables ($f\text{CO}_2$, DIC, TA, pH) facilitates calculation of the whole marine carbonate system. As will be shown later, the variable pH is not recommended to be the variable accompanying $f\text{CO}_2$, as the resulting uncertainty from the error propagation is too high. It is still important for FOS to measure pH, since it is vital for validating the $f\text{CO}_2$ data. This document presents the calculations for the whole carbonate system, including the error propagation coming from the variables themselves and from the constants used.

To estimate a meaningful accuracy for the measurement of a second carbon variable it was decided to follow the Global Ocean Observing System (GOOS) essential ocean variables (EOV's) and the Global Ocean Acidification Observing network (GOA-ON) approach of "weather" and "climate goals" for uncertainty calculation (Newton et al. 2014):

The "weather goal" is defined as measurements of quality sufficient to identify

- relative spatial patterns and short-term variations
- supporting mechanistic responses to and impact on local, immediate ocean acidification dynamics
- This implies an uncertainty of:

pH	~0.02
TA, DIC	~10 $\mu\text{mol/kg}$
$f\text{CO}_2$	~2.5% relative uncertainty

The "climate goal" is defined as measurements of quality sufficient to

- assess long-term trends with a defined level of confidence
- support detection of the long-term anthropogenically driven changes in hydrographic conditions and carbon chemistry over multi-decadal time scales
- This implies an uncertainty of:

pH	~0.003
TA, DIC	~2 $\mu\text{mol/kg}$
$f\text{CO}_2$	~0.5% relative uncertainty

According to the questions above, this document is divided in two sections:

A – This part investigates the sensitivity of one of the three calculated carbon parameters (DIC, TA and pH) based on the input variables ($f\text{CO}_2$ and DIC, TA, or pH). This is done separately for SOOP lines and FOS, as the quality requirements for these two station types are different.

B – This part investigates the resulting uncertainty of $f\text{CO}_2$ by using different pairs of input variables, which is important for validation of $f\text{CO}_2$ measurements at FOS.

All calculations were performed using the Matlab versions of CO2sys (van Heuven et al. 2009; Lewis and Wallace 1998; Orr et al. 2018) and the accompanied error calculations provided by Orr et al. (2015). The uncertainty of a variable is noted as $u(\text{variable name})$.

Please note: When using the Matlab version of the error propagation (errors.mat), be sure to update the file co2sys.mat provided by Orr et al. (2018), as there are minor changes compared to the version of van Heuven et al. (2009).

A – How accurately should the second carbon variable be measured on SOOP lines and at FOS?

The calculations in this section were done for different $f\text{CO}_2$ values between 250 and 700 μatm and for different temperatures between 5 and 25°C. TA and pH was calculated for each $f\text{CO}_2$ using a DIC value of 1950 $\mu\text{mol/kg}$ at 15°C. This should ensure a comparable carbonate system for further calculations.

%%%%%%%%%%%%%%%%%%%%%%%%%%%%%%%%%%%%%%%%%%%%%%%%%%%%%%%%%%%%%%%%%%%%%%%%%

$f\text{CO}_2$ = [250, 300, 350, 400, 450, 500, 550, 600, 700] μatm
DIC = 1950 $\mu\text{mol/kg}$
Temperature = [5, 10, 15, 20, 25] °C
Salinity = 35; in pressure = out pressure = 5 dbar; SI = 0; PO4 = 0;
pH scale: total scale
K1K2 constants: (Lueker et al., 2000)
KSO4 constants: (Dickson, 1990)
TB (total boron): (Uppström, 1974)

Uncertainties in constants:

$u(\text{pK0}) = 0.002$
 $u(\text{pK1}) = 0.0055$
 $u(\text{pK2}) = 0.01$
 $u(\text{pKb}) = 0.01$
 $u(\text{pKw}) = 0.01$
 $u(\text{pKspa}) = 0.02$
 $u(\text{pKspc}) = 0.02$
 $u(\text{Boron}) = 0.02$

Uncertainties in variables:

Salinity: $u(S) = 0.1$
Temperature: $u(T) = 0.05$

%%%%%%%%%%%%%%%%%%%%%%%%%%%%%%%%%%%%%%%%%%%%%%%%%%%%%%%%%%%%%%%%%%%%%%%%%

SOOP lines

ICOS SOOP lines measure sea surface $f\text{CO}_2$ with an uncertainty of $2 \mu\text{atm}$ ($u(f\text{CO}_2) = 2 \mu\text{atm}$). For the other 3 variables, standard uncertainties were used that can be reached by well-equipped marine carbon labs.

$$u(f\text{CO}_2) = 2 \mu\text{atm}, u(\text{DIC}) = u(\text{TA}) = 2 \mu\text{mol/kg}; u(\text{pH}) = 0.001$$

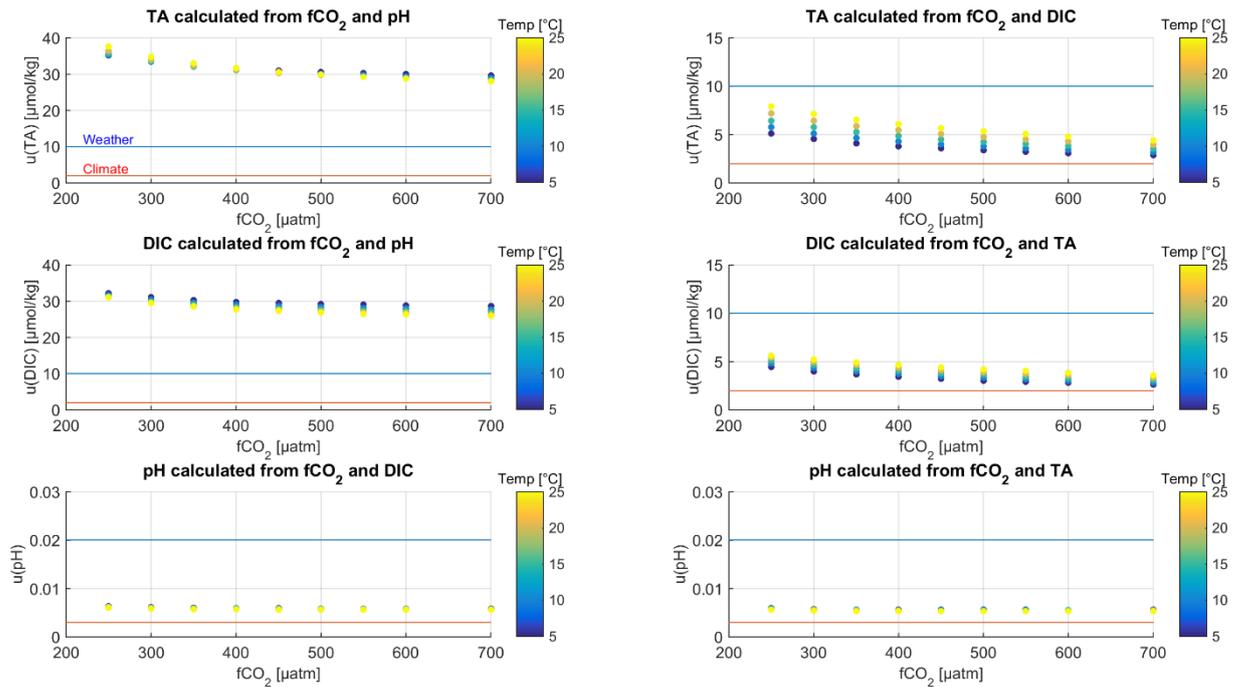


Figure 1a: Uncertainty of calculated variables of the marine carbonate system using $f\text{CO}_2$ and a second carbon variable. Standard uncertainties ($u(f\text{CO}_2) = 2 \mu\text{atm}$, $u(\text{DIC}) = u(\text{TA}) = 2 \mu\text{mol/kg}$; $u(\text{pH}) = 0.001$) were used for this calculation. The blue and red lines denote the “weather” and “climate goal”, respectively, for each variable according to Newton et al. (2014).

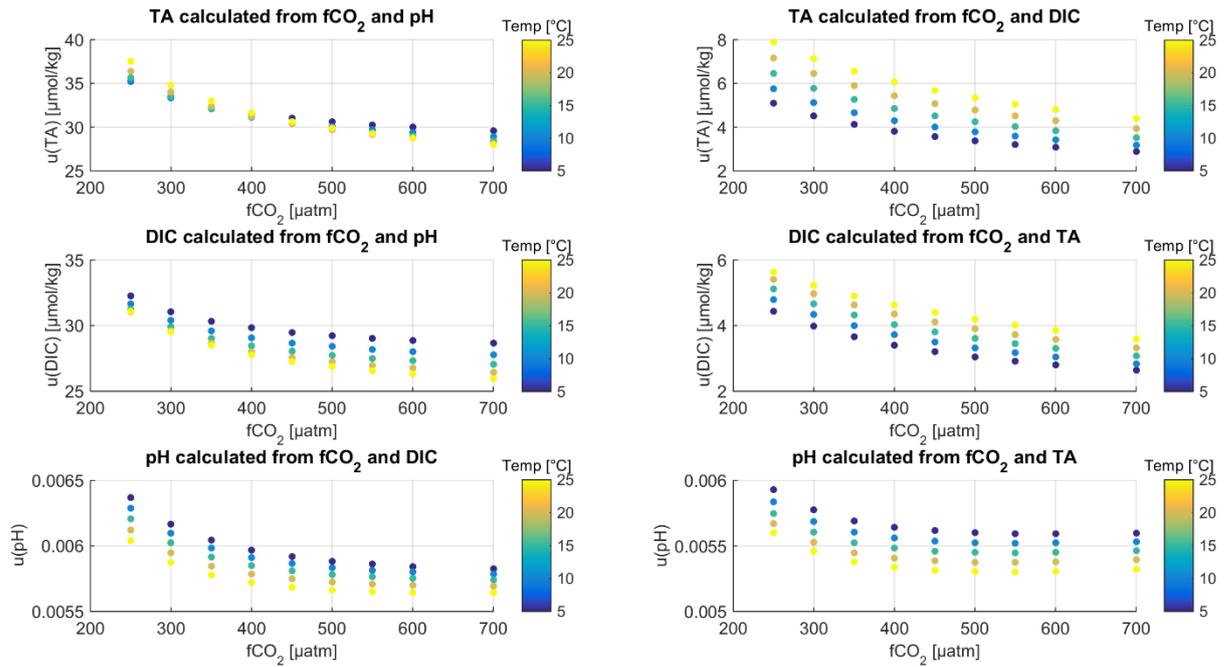


Figure 1b: Same as Figure 1a but without the lines for the “weather” and “climate goal” for better scaling.

Figure 1a shows that when $f\text{CO}_2$ and any second carbon variable are measured, one will never meet the thresholds for the “climate goal” when calculating the other two variables. Following the definition of the “weather goal” instead, the uncertainties of the second carbon variable were adjusted (Figures 2 and 3), so that the calculated variables met the “weather goal”. When using lower uncertainties for the input variables the resulting uncertainty of the calculated variables will be better. In Figure 2 and 3, the maximum uncertainty for the second input variable (beside $f\text{CO}_2$) was set to result in uncertainty estimates for the calculated variables within the “weather goal”. When using $f\text{CO}_2$ and pH, none of the resulting uncertainties fulfills the “weather goal”. Thus, pH and $f\text{CO}_2$ cannot be used for any calculations of the marine carbonate system for ICOS quality assessments. As mentioned above, nonetheless, pH is included in the following calculations in order to capture the entire system.

Input variables: $f\text{CO}_2$ and DIC

The output from the calculations shown in Figure 1 showed, that with an uncertainty for $f\text{CO}_2$ of 2 μatm , the uncertainty of DIC can increase to 5 $\mu\text{mol}/\text{kg}$ and the uncertainty for calculated TA and pH are below the thresholds for the “weather goal” of 10 $\mu\text{mol}/\text{kg}$ and 0.02 pH units, respectively.

$u(f\text{CO}_2) = 2 \mu\text{atm}$, $u(\text{DIC}) = 5 \mu\text{mol}/\text{kg}$

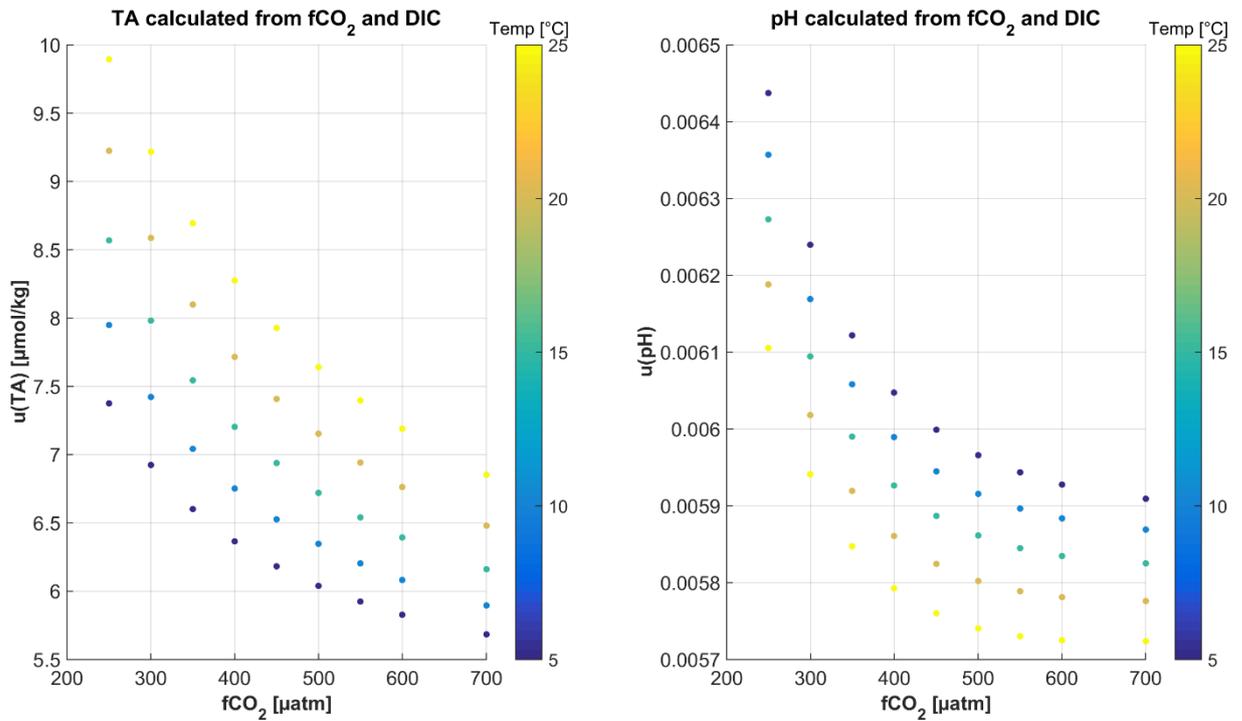


Figure 2: Uncertainty of TA and pH calculated from $f\text{CO}_2$ and DIC. With the assigned uncertainty of 5 $\mu\text{mol}/\text{kg}$ for DIC, TA and pH can be calculated to meet the “weather goal”.

Input variables: $f\text{CO}_2$ and TA

The output from the calculations shown in Figure 1 indicates, that with an uncertainty for $f\text{CO}_2$ of $2 \mu\text{atm}$, the uncertainty of TA can increase to $10 \mu\text{mol/kg}$ and the uncertainty for calculated DIC and pH are below the thresholds for the “weather goal” of $10 \mu\text{mol/kg}$ and 0.02 pH units, respectively.

$$u(f\text{CO}_2) = 2 \mu\text{atm}, u(\text{TA}) = 10 \mu\text{mol/kg}$$

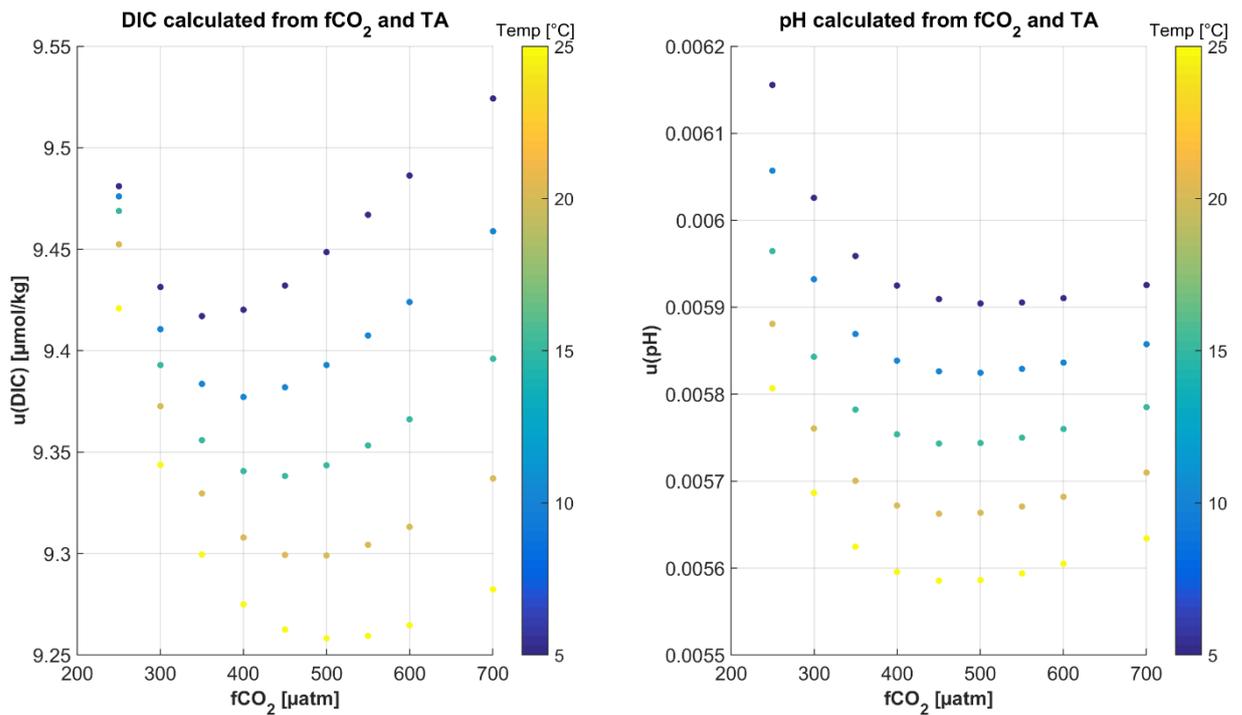


Figure 3: Uncertainty of DIC and pH calculated from $f\text{CO}_2$ and TA. With the assigned uncertainty of $10 \mu\text{mol/kg}$ for TA, DIC and pH can be calculated to meet the “weather goal”.

Figure 1a also shows that if one is aiming for the “climate goal”, this will not be reached by using $f\text{CO}_2$ and any other carbon parameter. The “climate goal” can only be reached with measured data. It also shows that using pH together with $f\text{CO}_2$ always leads to results with uncertainties higher than even the “weather goal”. As mentioned above, pH and $f\text{CO}_2$ should not be used for any calculations of the marine carbonate system for ICOS quality assessments. Table 2 summarizes the findings from above.

Table 2: “Weather goal” SOOP lines: Resulting uncertainties depending on the chosen input uncertainties. Green fields mean that the “weather goal” is met, red fields mean that the uncertainty is too high to meet the “weather goal”. The uncertainty of $f\text{CO}_2$ is always $2 \mu\text{atm}$.

$f\text{CO}_2$ + one of the variables below	Input uncertainty of input variable	DIC [$\mu\text{mol/kg}$]	pH	TA [$\mu\text{mol/kg}$]
DIC	$5 \mu\text{mol/kg}$		<<0.02	<10
pH	0.001	>25		>27
TA	$10 \mu\text{mol/kg}$	<10	<<0.02	

FOS

ICOS FOS measure sea surface $f\text{CO}_2$ with an uncertainty of $10 \mu\text{atm}$ ($u(f\text{CO}_2) = 10 \mu\text{atm}$). For the other 3 variables, standard uncertainties were used that can be reached by well-equipped marine carbon labs, as for the SOOP lines.

$u(f\text{CO}_2) = 10 \mu\text{atm}$, $u(\text{DIC}) = u(\text{TA}) = 2 \mu\text{mol/kg}$, $u(\text{pH}) = 0.001$

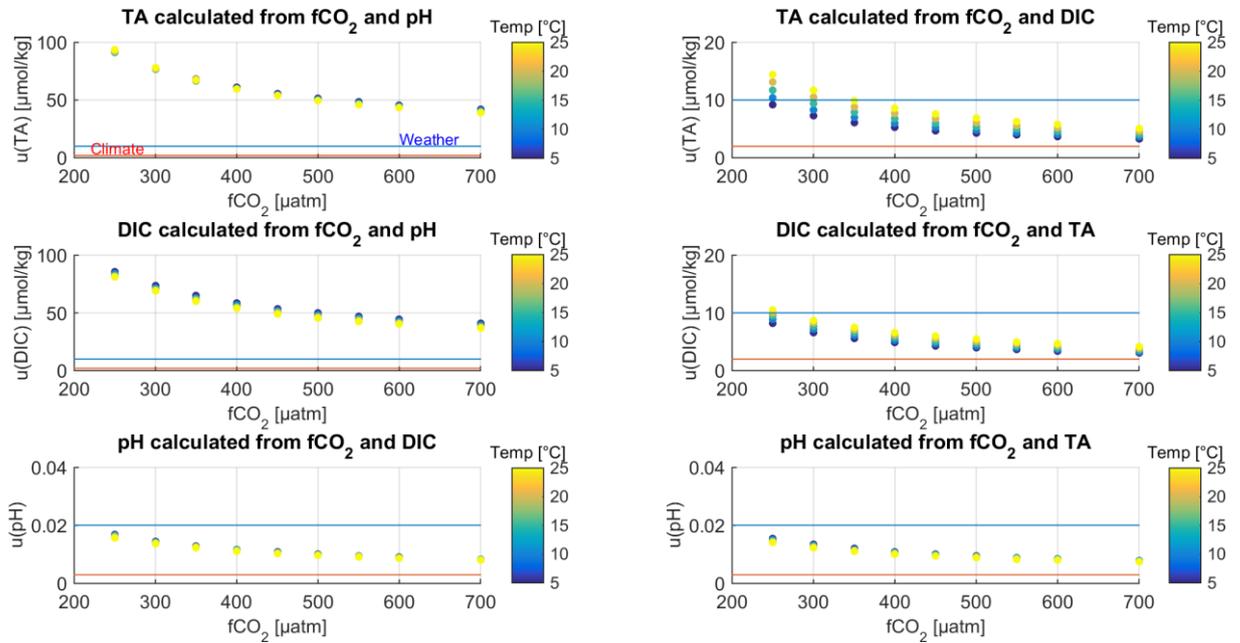


Figure 4a: Uncertainty of calculated variables of the marine carbonate system using $f\text{CO}_2$ and a second carbon variable. Standard uncertainties ($u(f\text{CO}_2) = 10 \mu\text{atm}$, $u(\text{DIC}) = u(\text{TA}) = 2 \mu\text{mol/kg}$; $u(\text{pH}) = 0.001$) were used for this calculation. The blue and red lines denote the “weather” and “climate goal”, respectively, for each variable according to Newton et al. (2014).

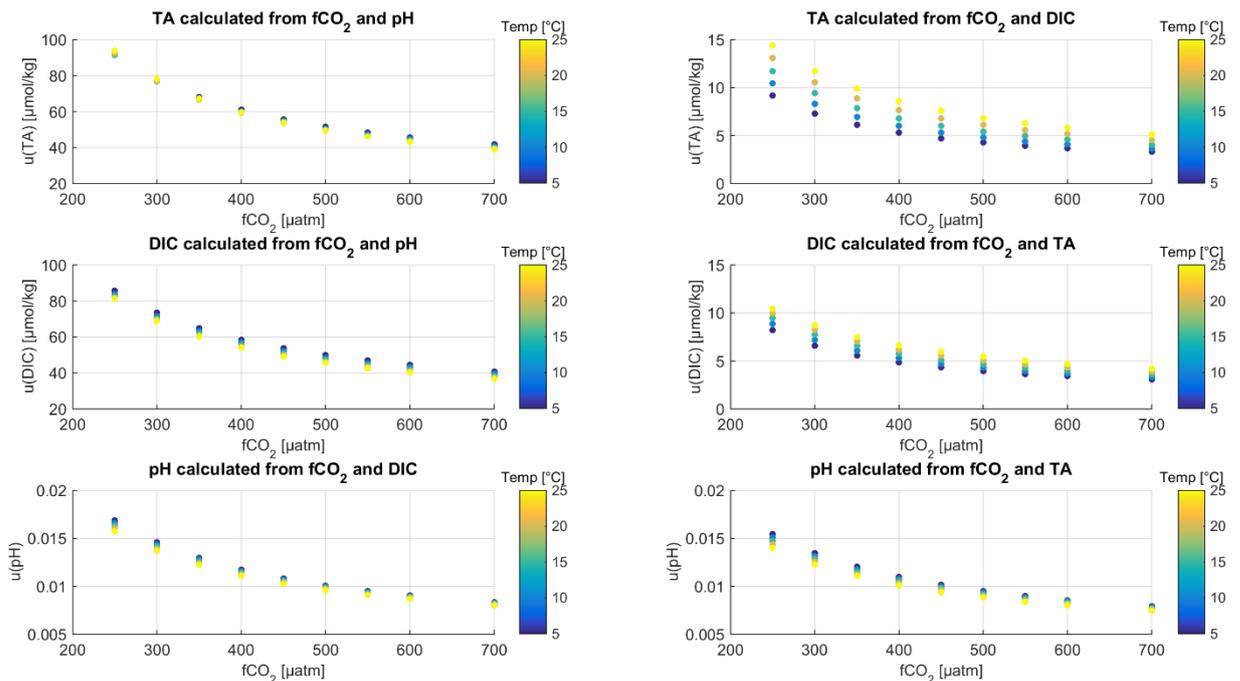


Figure 4b: Same as Figure 4a but without the lines for the “weather” and “climate goal” for better scaling.

Figure 4a shows that when $f\text{CO}_2$ (with $u(f\text{CO}_2)=10 \mu\text{atm}$) and any second carbon variable are measured, one will never meet the thresholds for the “climate goal” when calculating the other two variables. Following the definition of the “weather goal” instead, the uncertainties of the second carbon variable were adjusted (Figures 5 and 6), so that the calculated variables met the “weather goal”. In Figure 5 and 6, the maximum uncertainty for the second input variable (beside $f\text{CO}_2$) was set to result in uncertainty estimates for the calculated variables within the “weather goal”. When using $f\text{CO}_2$ and pH, none of the resulting uncertainties fulfills the “weather goal”. Thus, pH and $f\text{CO}_2$ cannot be used for any calculations of the marine carbonate system for ICOS quality assessments. As mentioned above, nonetheless, pH is included in the following calculations in order to capture the entire system.

Input variables: $f\text{CO}_2$ and DIC

The output from the calculations shown in Figure 4b showed that with an uncertainty for $f\text{CO}_2$ of $10 \mu\text{atm}$, the uncertainty of DIC must be as low as possible ($2 \mu\text{mol/kg}$) and the uncertainty for calculated alkalinity and pH are below the thresholds for the “weather goal” of $10 \mu\text{mol/kg}$ (only for $f\text{CO}_2$ values greater than $350 \mu\text{atm}$) and 0.02 pH units , respectively.

$u(f\text{CO}_2) = 10 \mu\text{atm}$, $u(\text{DIC}) = 2 \mu\text{mol/kg}$

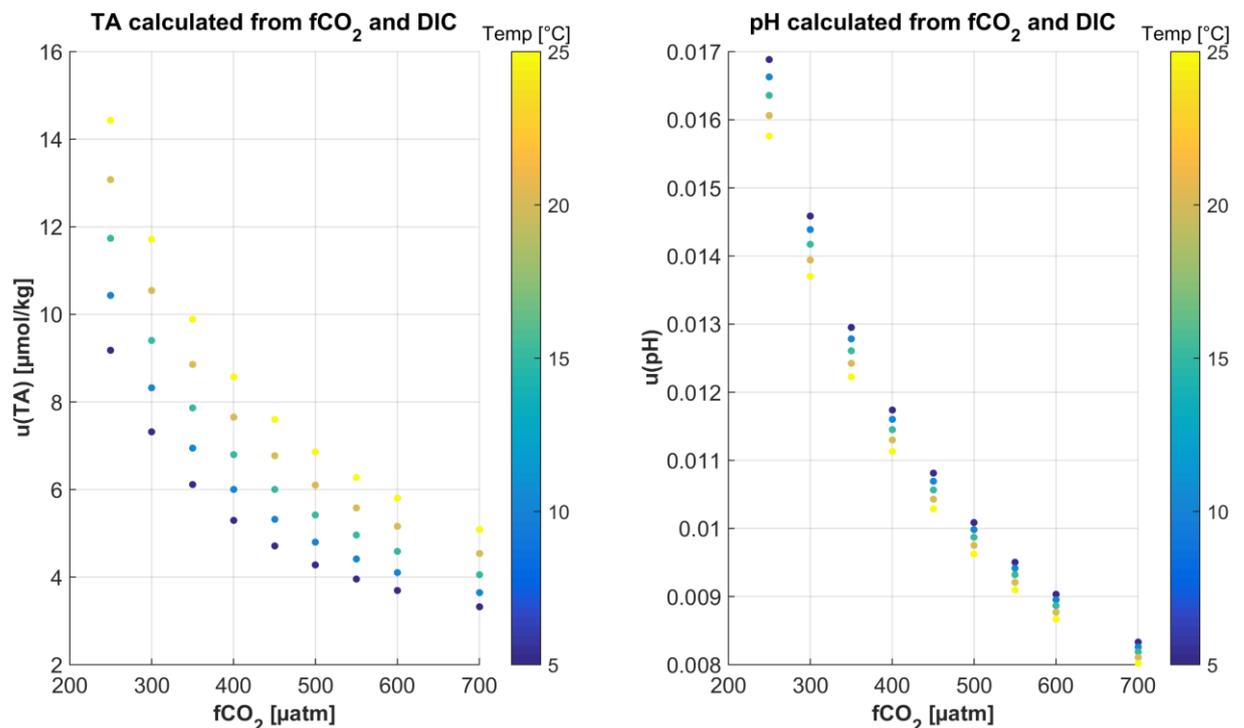


Figure 5: Uncertainty of TA and pH calculated from $f\text{CO}_2$ and DIC. With the assigned uncertainty of $2 \mu\text{mol/kg}$ for DIC, TA (for $f\text{CO}_2 > 350 \mu\text{atm}$) and pH can be calculated to meet the “weather goal”.

Input variables: $f\text{CO}_2$ and TA

The output from the calculations shown in Figure 4b showed that with an uncertainty for $f\text{CO}_2$ of 10 μatm the uncertainty of TA can increase to 4 $\mu\text{mol/kg}$ and the uncertainty for calculated DIC and pH are below the thresholds for the “weather goal” of 10 $\mu\text{mol/kg}$ (only for $f\text{CO}_2$ values greater than 250 μatm) and 0.02 pH units, respectively.

$u(f\text{CO}_2) = 10 \mu\text{atm}$, $u(\text{TA}) = 4 \mu\text{mol/kg}$

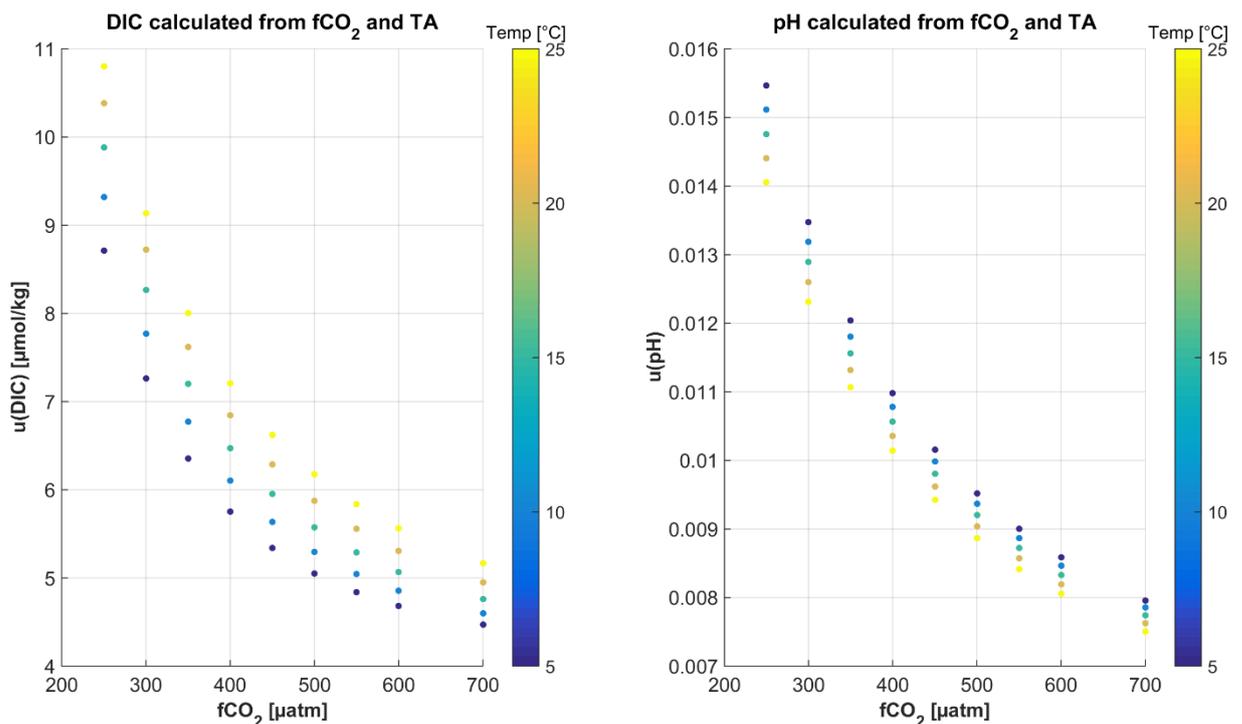


Figure 6: Uncertainty of DIC and pH calculated from $f\text{CO}_2$ and TA. With the assigned uncertainty of 4 $\mu\text{mol/kg}$ for TA, DIC (for $f\text{CO}_2 > 300 \mu\text{atm}$) and pH can be calculated to meet the “weather goal”.

Figure 4a also shows that if one is aiming for the “climate goal”, this will not be reached by using $f\text{CO}_2$ and any other carbon parameter. The “climate goal” can be only be reached with measured data. It also shows that using pH together with $f\text{CO}_2$ always leads to results higher than the “weather goal”. As mentioned above for SOOP lines, pH and $f\text{CO}_2$ should not be used for any calculations of the marine carbonate system for ICOS quality assessments. Table 3 summarizes the findings from above.

Table 3: “Weather goal” FOS: Resulting uncertainties depending on the chosen input uncertainties. Green fields mean that the “weather goal” is met, red fields mean that the uncertainty is too high to meet the “weather goal”. The uncertainty of $f\text{CO}_2$ is always 10 μatm .

$f\text{CO}_2$ + one of the variables below	Input uncertainty of input variable	DIC [$\mu\text{mol/kg}$]	pH	TA [$\mu\text{mol/kg}$]
DIC	2 $\mu\text{mol/kg}$		<<0.02	<10*
pH	0.001	>30		>30
TA	4 $\mu\text{mol/kg}$	<10*	<<0.02	

*for $f\text{CO}_2 > 300 \mu\text{atm}$

- **Concluding part A**

When aiming for uncertainties of marine carbon variables that fulfill the “climate goal” as defined by Newton et al. (2014), the carbon variables need to be measured. Due to the error propagation, the resulting uncertainty when calculating variables from $f\text{CO}_2$ and another variable is always above the threshold of $2 \mu\text{mol/kg}$ for DIC and TA and 0.003 for pH.

It was shown that using $f\text{CO}_2$ and pH will always result in uncertainties higher than the uncertainty limits stated for the “weather goal” ($10 \mu\text{mol/kg}$ for DIC and TA, 0.02 for pH).

Figures 2 and 3 and Table 1 show that SOOP lines need to provide the additional variable DIC with an uncertainty of $5 \mu\text{mol/kg}$ or TA with an uncertainty of $10 \mu\text{mol/kg}$, in order to become a Class 1 station.

Figures 5 and 6 and Table 2 show that FOS need to provide the additional variable DIC with an uncertainty of $2 \mu\text{mol/kg}$ or TA with an uncertainty of $4 \mu\text{mol/kg}$, in order to become a Class 1 station.

B - Can $f\text{CO}_2$ be calculated with a sufficient uncertainty by two other carbon parameters?

The calculations in this section were done for four different $f\text{CO}_2$ levels between 250 and 600 μatm and for different temperatures between 5 and 25°C. TA and pH was calculated for each $f\text{CO}_2$ using a constant DIC value of 1950 $\mu\text{mol/kg}$ at 15°C. This should ensure a consistent carbonate system for further calculations.

For each pair of input variables, four different uncertainties were used for the calculation of $f\text{CO}_2$:

%%%%%%%%%%%%%%%%%%%%%%%%%%%%%%%%%%%%%%%%%%%%%%%%%%%%%%%%%%%%%%%%%%%%%%%%%

$f\text{CO}_2$ = [250, 350, 450, 600] μatm
DIC = 1950 $\mu\text{mol/kg}$
Temperature = [5, 10, 15, 20, 25] °C
Salinity = 35; in pressure = out pressure = 5 dbar; SI = 0; PO4 = 0;
pH scale: total scale
K1K2 constants: (Lueker et al., 2000)
KSO4 constants: (Dickson, 1990)
TB (total boron): (Uppström, 1974)

Uncertainties in constants:

$u(\text{pK0}) = 0.002$
 $u(\text{pK1}) = 0.0055$
 $u(\text{pK2}) = 0.01$
 $u(\text{pKb}) = 0.01$
 $u(\text{pKw}) = 0.01$
 $u(\text{pKspa}) = 0.02$
 $u(\text{pKspc}) = 0.02$
 $u(\text{Boron}) = 0.02$

Uncertainties in variables:

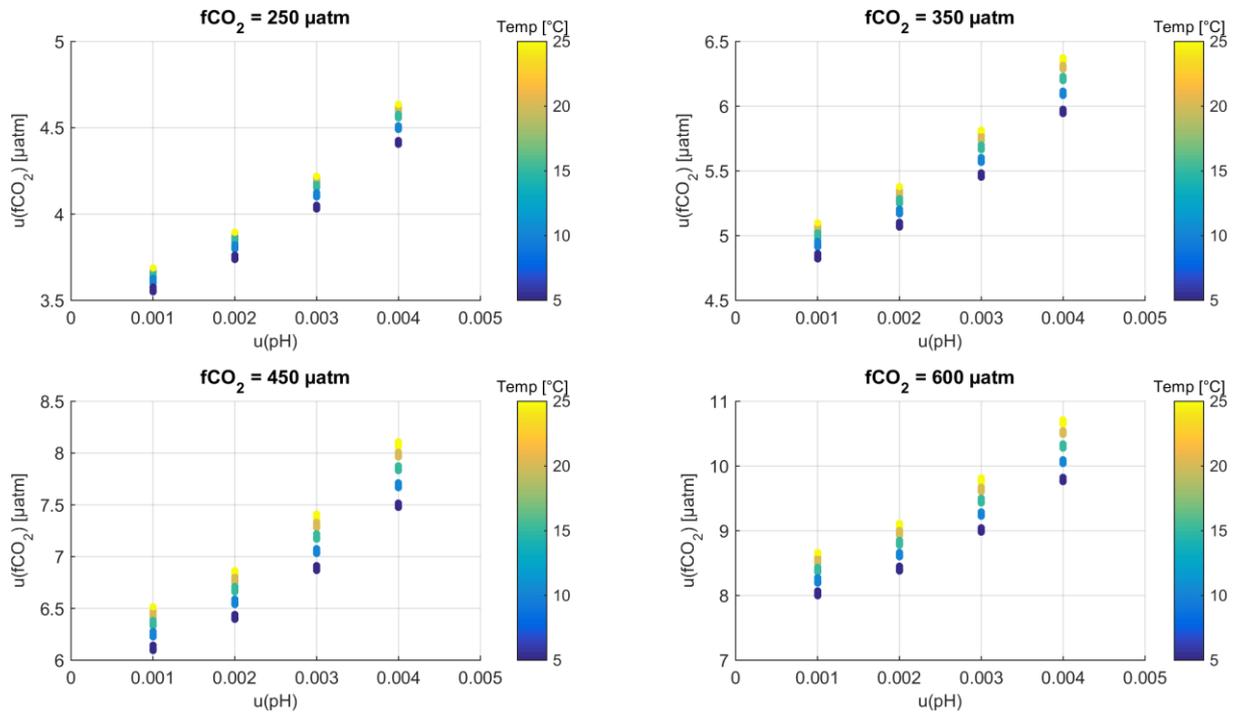
Salinity: $u(S) = 0.1$
Temperature: $u(T) = 0.005$
 $u(\text{pH}) = [0.001, 0.002, 0.003, 0.004]$
 $u(\text{TA}) = [1, 2, 3, 4]$
 $u(\text{DIC}) = [1, 2, 3, 4]$

%%%%%%%%%%%%%%%%%%%%%%%%%%%%%%%%%%%%%%%%%%%%%%%%%%%%%%%%%%%%%%%%%%%%%%%%%

The following figures show the resulting uncertainty in $f\text{CO}_2$ ($u(f\text{CO}_2)$) by using two variables with different uncertainties. Each figure shows the results at four different $f\text{CO}_2$ levels and different temperatures. The top part of the figures shows the $u(f\text{CO}_2)$ plotted as the uncertainty of the first input variable and the bottom part shows the same $u(f\text{CO}_2)$ plotted versus the uncertainties of the second input variable.

Using TA and pH as input variables:

$u(fCO_2)$ vs. $u(pH)$



$u(fCO_2)$ vs. $u(TA)$

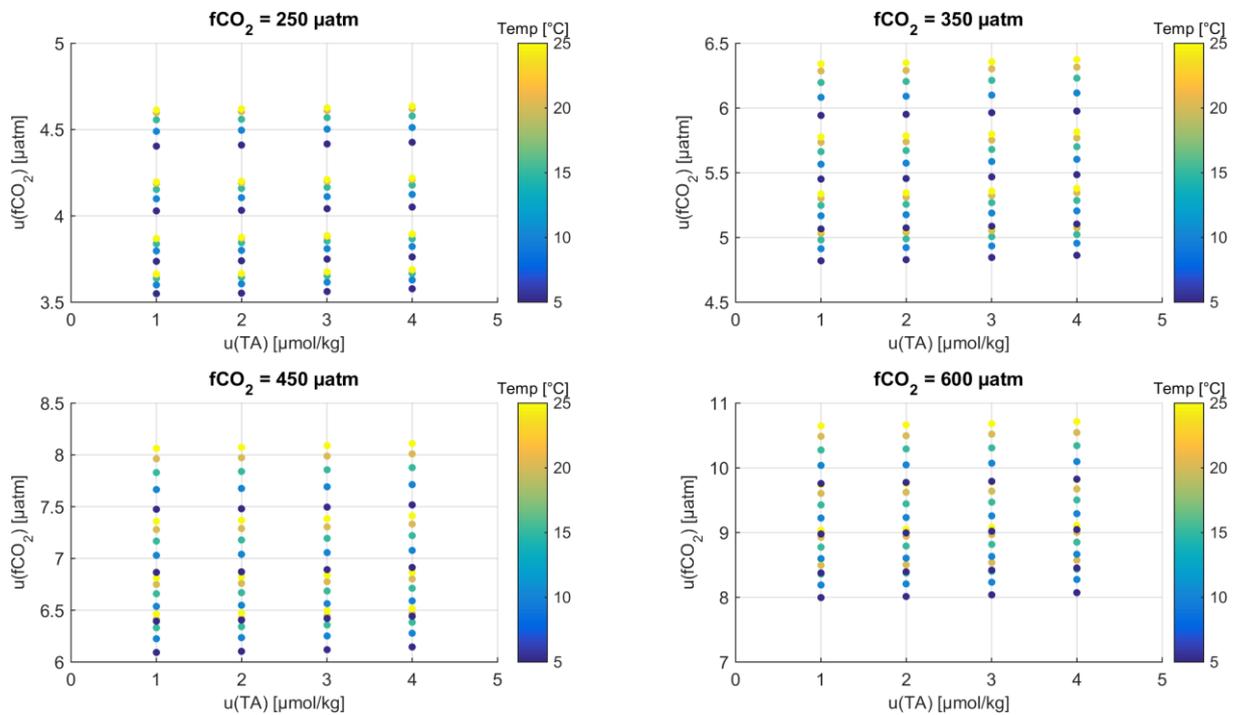
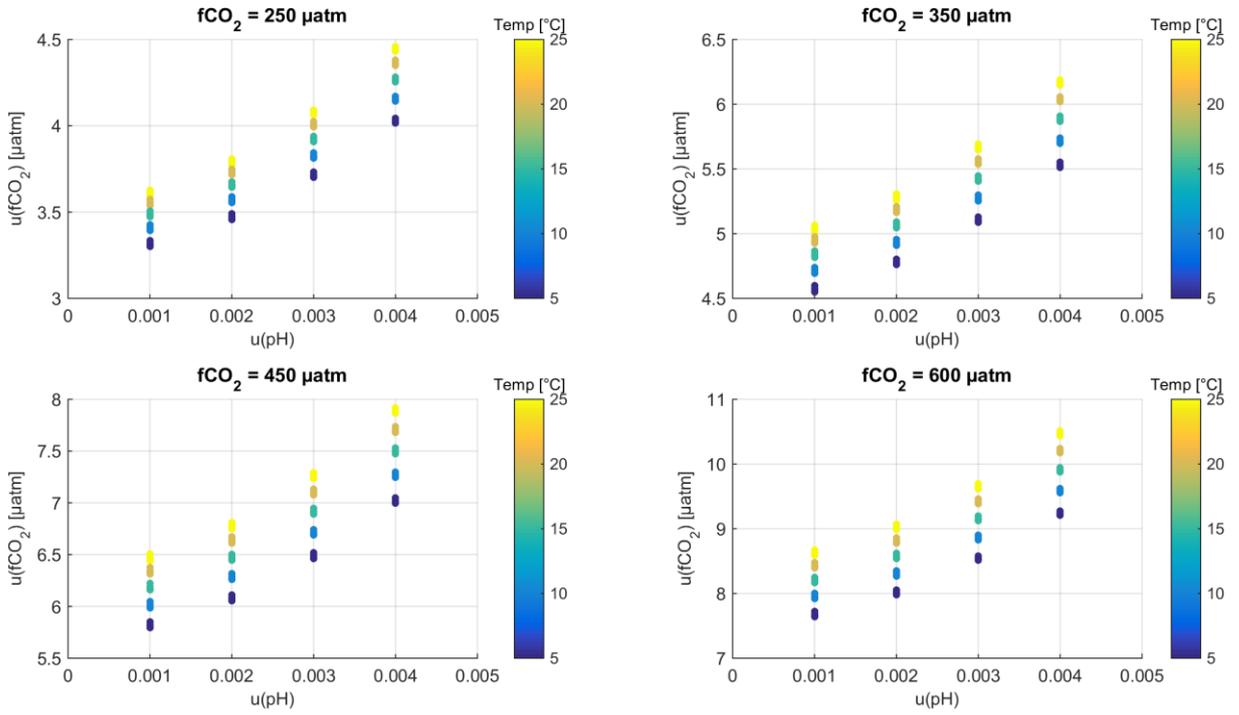


Figure 7: Uncertainty of fCO_2 calculated from TA and pH for four different fCO_2 levels and different temperatures. The upper four panels show the impact of the uncertainty of pH ($u(pH)$) and the lower panels show the impact of $u(TA)$.

Using DIC and pH as input variables:

$u(fCO_2)$ vs. $u(pH)$



$u(fCO_2)$ vs. $u(DIC)$

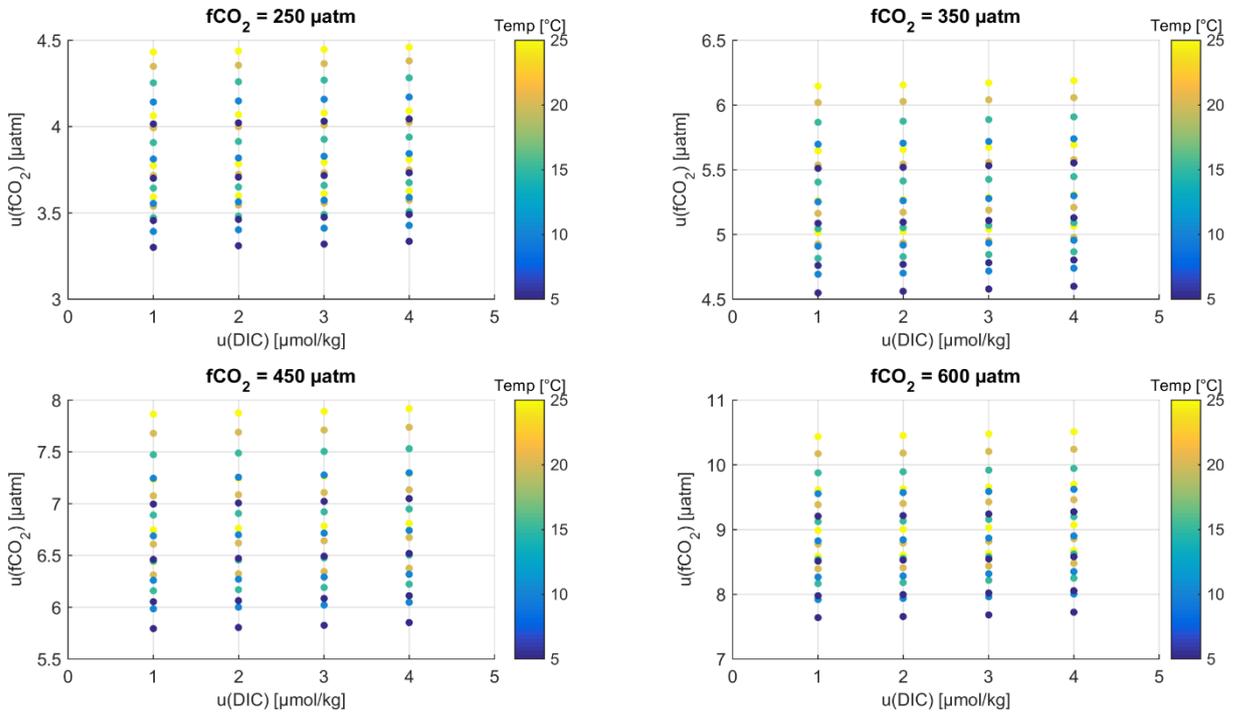
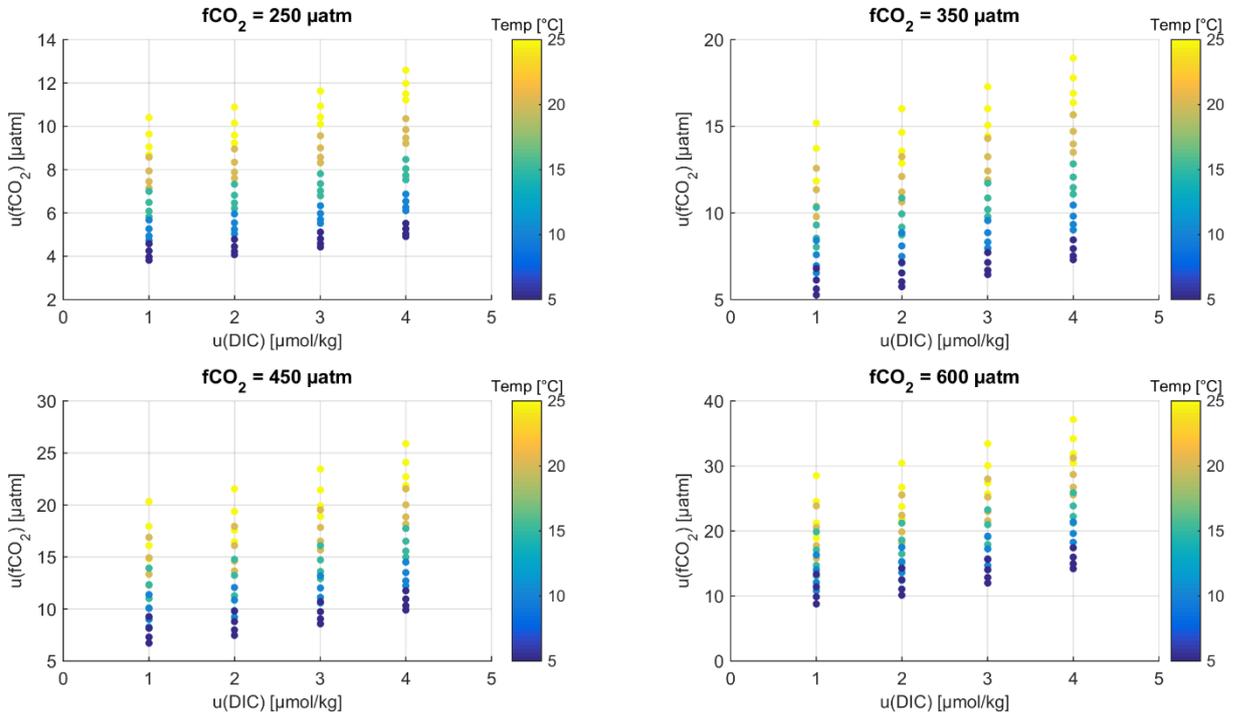


Figure 8: Uncertainty of fCO_2 calculated from DIC and pH for four different fCO_2 levels and different temperatures. The upper four panels show the impact of $u(pH)$ and the lower panels show the impact of $u(DIC)$.

Using DIC and TA as input variables:

$u(fCO_2)$ vs. $u(DIC)$



$u(fCO_2)$ vs. $u(TA)$

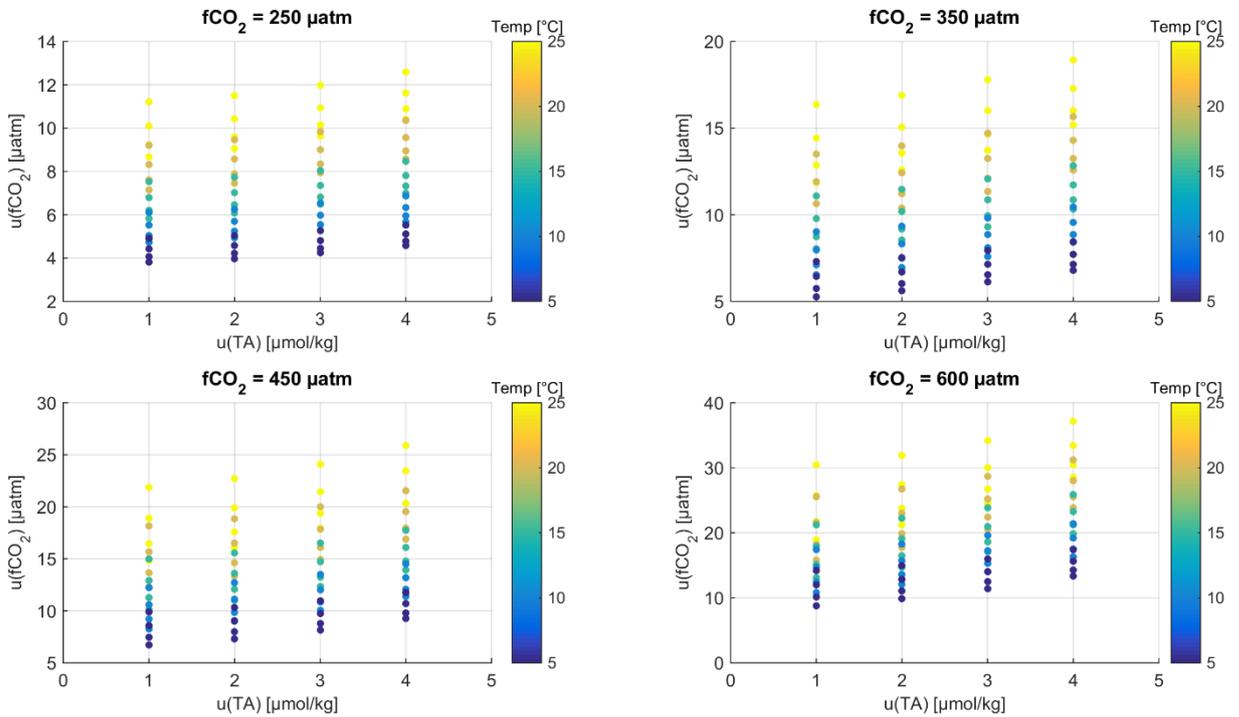


Figure 9: Uncertainty of fCO_2 calculated from DIC and TA for four different fCO_2 levels and different temperatures. The upper four panels show the impact of $u(DIC)$ and the lower panels show the impact of $u(TA)$.

Concluding part B

- Calculating $f\text{CO}_2$ from two carbonate variables is only suitable for FOS as there is no combination that results in an uncertainty for calculated $f\text{CO}_2$ better than $3.5 \mu\text{atm}$.
- Using TA and pH: $f\text{CO}_2$ can be calculated with an uncertainty better than $10 \mu\text{atm}$ for $f\text{CO}_2$ values below ca. $600 \mu\text{atm}$ when $u(\text{pH}) < 0.003$ and $u(\text{TA}) < 4 \mu\text{mol/kg}$.
- Using DIC and pH: $f\text{CO}_2$ can be calculated with an uncertainty better than $10 \mu\text{atm}$ for $f\text{CO}_2$ values below ca. $600 \mu\text{atm}$ when $u(\text{pH}) < 0.003$ and $u(\text{DIC}) < 4 \mu\text{mol/kg}$.
- DIC and TA can only be used for water temperatures below 15°C and $f\text{CO}_2$ levels below $450 \mu\text{atm}$.

Summary

When calculating carbon variables using CO2sys, the error propagation should always be included. When using the error propagation code from Orr et al. (2018), one also needs to update the CO2sys files as they also include minor changes to account for the error propagation.

The labelling scheme of ICOS-Oceans, as agreed during the Monitoring Station Assembly (MSA) meeting in Southampton in 2019, defines two goals with respect to the marine carbon cycle:

- Quantifying air-sea CO₂ fluxes
- Assessing the variability and drivers of these fluxes

Limitations of ocean *f*CO₂ measurement capabilities at FOS mean that fluxes cannot be determined to the accuracy desired for large scale carbon budgets. Meanwhile, it is most common that SOOP lines measure *f*CO₂ rather than other carbon system variables. Therefore, *f*CO₂ needs to be measured with an accuracy of 2 μatm on SOOP lines and 10 μatm at FOS. To estimate a meaningful accuracy for the measurement of a second carbon variable it was decided to use the definitions of the “weather goal” and the “climate goal” as defined by Newton et al. (2014) for the Global Ocean Acidification Observing Network (GOA-ON).

One important outcome is that when aiming for data with uncertainties fulfilling the “climate goal” these data need to be measured. There is no combination of *f*CO₂ and a second carbon variable that allow the calculation of the other two variables with sufficient uncertainty. Using *f*CO₂ and pH will result in uncertainties that are above the thresholds for the “weather goal”. Using the results from part A, uncertainty estimates can be added to the variables in Table 1, so that it can be rewritten as shown in Table 4:

		SOOP lines	FOS
Class 2 (minimum variables)	required	<i>f</i> CO ₂ (±2 μatm)	<i>f</i> CO ₂ (±10 μatm) Alkalinity (±4 μmol/kg) or DIC (±2 μmol/kg) Oxygen
Class 1 (additional variables)		Alkalinity (±10 μmol/kg) or DIC (±5 μmol/kg) Oxygen	Surface: Nutrients (nitrate, silicate and phosphate)

Part B investigated which variables can be used and to what uncertainty they need to be measured to evaluate *f*CO₂ measurements at ICOS stations. Using two variables of the marine carbonate system to calculate *f*CO₂ is only suitable for FOS. There is no combination that results in an uncertainty for calculated *f*CO₂ better than 3.5 μatm. Of course, the results depend heavily on temperature and often on the *f*CO₂ level itself. When evaluating a FOS, the results need to be analyzed in detail for each station.

When using TA and pH, *f*CO₂ can be calculated with an uncertainty better than 10 μatm for *f*CO₂ values below ca. 600 μatm using *u*(pH)<0.003 and *u*(TA)<4 μmol/kg. The uncertainty of pH has the larger effect. The *f*CO₂ can be calculated with an uncertainty of below 5 μatm at low *f*CO₂ levels (<350 μatm) and temperatures below 15°C.

When using DIC and pH, $f\text{CO}_2$ can be calculated with an uncertainty better than 10 μatm for $f\text{CO}_2$ values below ca. 600 μatm using $u(\text{pH}) < 0.003$ and $u(\text{DIC}) < 4 \mu\text{mol/kg}$. Again, the uncertainty of pH has the larger effect. The $f\text{CO}_2$ can be calculated with an uncertainty of below 5 μatm at low $f\text{CO}_2$ levels ($< 350 \mu\text{atm}$) and temperatures below 20°C.

When using DIC and TA, $f\text{CO}_2$ can be calculated with an uncertainty better than 10 μatm only for water temperatures below 15°C and $f\text{CO}_2$ levels below 450 μatm . The pair DIC-TA should not be used for evaluating ICOS FOS, as it gives sufficient results only in a limited temperature range.

References

van Heuven, S., D. Pierrot, E. Lewis, and D. W. R. Wallace. 2009. "MATLAB Program Developed for CO2 System Calculations. ORNL/CDIAC-105b." ORNL/CDIAC.

Lewis, E. and Douglas W. R. Wallace. 1998. "Program Developed for CO2 System Calculations." ORNL/CDIAC-105. Carbon Dioxide Information Analysis Center, Oak Ridge National Laboratory, U.S. Department of Energy, Oak Ridge, Tennessee.

Newton, J. A., R. A. Feely, E. B. Jewett, P. Williamson, and J. Mathis. 2014. "Global Ocean Acidification Observing Network: Requirements and Governance Plan." (September):57 pp.

Orr, J. C., J. M. Epitalon, and J. P. Gattuso. 2015. "Comparison of Ten Packages That Compute Ocean Carbonate Chemistry." *Biogeosciences* 12(5):1483–1510.

Orr, James C., Jean Marie Epitalon, Andrew G. Dickson, and Jean Pierre Gattuso. 2018. "Routine Uncertainty Propagation for the Marine Carbon Dioxide System." *Marine Chemistry* 207(June):84–107.

Uppström, Leif R. 1974. "The Boron/Chlorinity Ratio of Deep-Sea Water from the Pacific Ocean." *Deep Sea Res. Oceanogr. Abstr.* 21:161–62.